Natural Product Synthesis

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Enantioselective, Protecting-Group-Free Total Synthesis of Sarpagine Alkaloids—A Generalized Approach**

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Dedicated to Professor Johann Mulzer on the occasion of his 70th birthday

Abstract: A generalized synthetic access to sarpagine alkaloids through a joint synthetic sequence has been accomplished. Its applicability is showcased by the enantioselective total syntheses of vellosimine (1), N-methylvellosimine (3), and 10methoxyvellosimine (8). The synthetic sequence is concise (eight steps) from known compound 13, and requires no protecting groups. The indole heterocycle was introduced in the last step. This strategy allows access to sarpagine alkaloids through a shared synthetic route leading to precursor 10, which we term "privileged intermediate". Starting from this intermediate, all sarpagine alkaloids can be synthesized using phenylhydrazines with different substitution patterns (15–17). Our approach brings about the advantage, that synthesis optimization only needs to be performed once for many natural products. The key features of the synthesis are a [5+2]cycloaddition and a ring enlargement.

Sarpagine alkaloids belong to the group of monoterpenoid indole alkaloids. Their family consists of more than 90 congeners, which were mainly isolated from the plant family Apocynaceae (specifically from the genus Rauvolfia).[1] Cook et al. accomplished very elegant total syntheses of some sarpagines using Pictet-Spengler type chemistry (early introduction of the indole core).[2] With regard to the biosynthesis of these complex molecular architectures, primary as well as secondary cyclizations are well investigated by Stöckigt et al.[3] Yet, the biological potential of these beautiful architectures, especially when it comes to the investigation of synthetic analogues, is by far underexplored. Substructures of sarpagines were synthesized by Waldmann et al. for library design, and revealed potent tyrosine kinase inhibitors.^[4] Unfortunately the scarcity of the natural products themselves, and the lack of synthetic analogues of comparable complexity have so far hampered further biological investigations.

When we launched our synthetic program on sarpagine alkaloids (1–9), we not only aimed at solving the problem of material supply for a single family congener, but at accomplishing a joint synthetic sequence that enables access to a large number of sarpagines through a shared late-stage synthetic intermediate, which we term "privileged intermediate". At the outset we had to define the molecular structure of this intermediate. Therefore, the individual congeners were analyzed for common structure patterns as well as differences in stereochemistry and oxidation states.

Important structural variations such as: 1) additional rings [gardnutine (6), Figure 1], 2) variability of the absolute

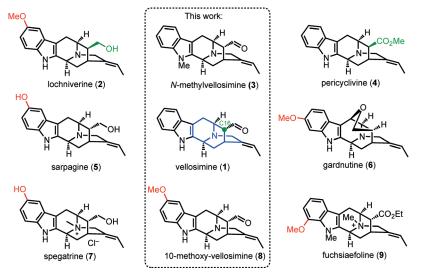


Figure 1. Selected family members of sarpagine alkaloids.

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configuration at C16 [biosynthetic numbering; highlighted in green in Figure 1], and 3) hydroxylation pattern of the indole core (highlighted in red in Figure 1) were identified. Furthermore, we assessed an octahydro-1*H*-2,6-methanoquinolizine system as common structural motif of all sarpagines (highlighted in blue in Figure 1 for vellosimine (1)).

With this analysis we directly deduced the retrosynthesis leading us to ketone 10, as the "privileged intermediate" (Figure 2 A). From this last synthetic intermediate all sarpagines can be synthesized in one step, introducing the different hydroxylation patterns of the respective natural products by



(+)-vellosimine (1)

A) Retrosynthetic analysis of (+)-vellosimine (1)

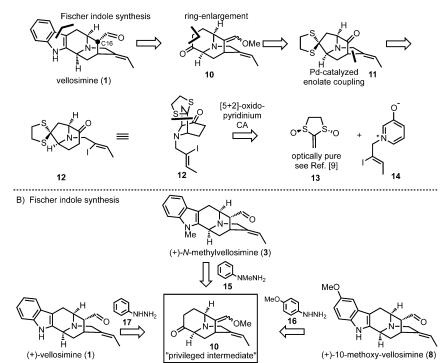


Figure 2. Retrosynthetic analysis for a generalized synthesis of sarpagines 1, 3, and 8.

using the corresponding phenylhydrazines (15–17) in a Fischer indole synthesis (Figure 2B).^[5] The absolute configuration of C16 (Figure 2A) can be adjusted to give the α -isomer, which is thermodynamically favored. [6] Ketone 10 itself was obtained through ring enlargement from compound 11.[7] The tricyclic system of 11 was formed through a palladiumcatalyzed enolate coupling[8] starting from bicyclic compound 12. This compound 12 represents the product of a [5+2]cycloaddition between oxidopyridinium ion 14 and Aggarwal's chiral ketene equivalent 13 (prepared in four steps).^[9] Similar pyridinium ions are known to undergo intermolecular

Scheme 1. The [5+2] cycloaddition reaction for the construction of the core structure. Reagents and conditions: a) NiPr₂Et, CH₂Cl₂, 12h, 77%; b) TFAA, NaI, MeCN, 0°C, 68%; c) L-selectride, THF, -78°C, 94%; d) KOtBu, PhOH, [Pd(PPh₃)₄], THF, reflux, 88%.

[5+2] cycloadditions with regioselectivities around 2:1, typically yielding the desired regioisomer 12 as a major product.[10] When we performed the cycloaddition of 13 and 14, we obtained our desired cycloadduct 19 in a 2:1 ratio (77 % combined yield) in favor of the desired regioisomer and 93% ee (Scheme 1). Luckily the stability of the vinyliodide moiety in 14 allowed the direct assembly of compound 19 without the need of protecting groups.

Reductive removal of the bis(sulfoxide) after the [5+2] cycloaddition was affected by treatment of 19 with trifluoroacetic anhydride and sodium iodide affording dithiolane 20 in 68% yield.[10] The next task was the formation of tricycle 11 through an intramolecular palladium-catalyzed enolate coupling. This was achieved in a sequential manner by treating 20 with L-selectride to form ketone 12, which was reacted with palladium (tetrakis)triphenylphosphine to yield tricyclic ketone 11 in 88% yield (Scheme 1).[7] This two-step process was also performed in one pot, but then the yields dropped significantly to about 50%.

Wittig reaction of 11 and deprotection of the dithiolane moiety with Meerwein's salt afforded ketone 21 (Scheme 2).[11] This set the stage for the ring enlargement reaction to give ketone 10 in 80 % yield. The ring enlargement proceeded smoothly and regioselectively, resulting in the exclusive insertion of the methylene group on the sterically less hindered side.[12]

(+)-10-methoxy-vellosimine (8)

With ketone 10 in hands we carried out the Fischer indole synthesis with phenylhydrazines 15–17.^[13] In the course of this reaction, enol ether 22 was converted to dimethylacetal 23, which we observed by ¹H NMR spectroscopy on the crude sample. 23 was hydrolyzed in situ affording (+)-vellosimine (1) (+)-N-methylvellosimine (3), and (+)-10methoxyvellosimine (8) in 52-63 % yield. As anticipated, the desired absolute configuration at C-16 was exclusively obtained under these reaction conditions in all three natural products 1, 3, and 8.[14]

An important feature of this synthesis is the strategy to pursue a "privileged intermediate", which paves the synthetic road to most of the sarpagine alkaloids in a single last transformation—in our case the Fischer indole synthesis. This avoids synthetic inconveniences normally observed when substituents have to be varied at the starting point of a synthesis, which at worst can cause (partial) redesign of the route.

In conclusion we have accomplished a generalized, enantioselective access to the sarpagine alkaloid family in a very concise manner. (+)-Vellosimine (1), (+)-N-methylvellosimine (3), and (+)-10-methoxyvellosimine (8) were synthesized in eight steps starting from known compounds

Scheme 2. Synthesis of intermediate **10**, and total syntheses of (+)-vellosimine (1), (+)-N-methylvellosimine (3), and (+)-10-methoxyvellosimine (8).Reagents and conditions: e) MeOCH=PPh₃; f) TFA, CH₂Cl₂, Me₃OBF₄, 58% (2 steps); g) TMSCH₂N₂, nBuLi, THF, then MeOH; then silica, 80%; h) AcCl, MeOH, Δ , then water.

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14 and 13 (12 steps from commercial materials) in 10–13% overall yields. The joint synthetic sequence requires no protecting groups, and can be carried out on multigram scale. Therefore the lack of synthetic analogues and the limited amount of material, currently hampering detailed biological testing, is no longer an issue. Our synthetic route will enable the production of almost every family member on demand. The syntheses of further sarpagine congeners are currently performed in our laboratories.

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3: R¹= H; R²= Me; 52%

8: R1= OMe; R2= H; 63%

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